

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
31 July 2003 (31.07.2003)

PCT

(10) International Publication Number
WO 03/063197 A1

(51) International Patent Classification⁷: **H01J 61/40**,
H01K 1/32, G02B 5/20, 5/22, C09D 183/04

(21) International Application Number: PCT/IB03/00084

(22) International Filing Date: 15 January 2003 (15.01.2003)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
02075301.8 24 January 2002 (24.01.2002) EP

(71) Applicant (for all designated States except US): **KONINKLIJKE PHILIPS ELECTRONICS N.V.** [NL/NL];
Groenewoudseweg 1, NL-5621 BA Eindhoven (NL).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **BOEHMER, Marcel, R.** [NL/NL]; Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL). **WAGEMANS, Melanie, M., H.** [NL/NL];
Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL).

(74) Agent: **VAN WERMESKERKEN, Stephanie, C.**; Internationaal Octrooibureau B.V., Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: LIGHT-TRANSMITTING SUBSTRATE PROVIDED WITH A LIGHT-ABSORBING COATING

(57) Abstract: Disclosed is a light-transmitting substrate which is at least partly provided with a light-absorbing coating. Said coating comprises light-absorbing particles that are incorporated in a sol-gel matrix. The light-absorbing particles comprise silver or gold or a mixture thereof, and the coating further comprises a dimethyl-aminosilane. Furthermore, an electric lamp is disclosed comprising a light-transmitting lamp vessel that accommodates a light source. Said lamp vessel comprises the above light-transmitting substrate. A method of preparing a light-absorbing coating to be applied to a light-transmitting substrate, and a method of applying a light-absorbing coating to a light-transmitting substrate are also disclosed.

WO 03/063197 A1

LIGHT-TRANSMITTING SUBSTRATE PROVIDED WITH A LIGHT-ABSORBING COATING

The present invention relates to a light-transmitting substrate that is at least partly provided with a light-absorbing coating, said coating comprising light-absorbing particles that are incorporated in a sol-gel matrix. The invention further relates to an electric lamp comprising a light-transmitting lamp vessel that accommodates a light source, wherein
5 said lamp vessel comprises the above light-transmitting substrate. Furthermore, the present invention relates to a method of preparing a light-absorbing coating to be applied to a light-transmitting substrate, as well as a method of applying a light-absorbing coating to a light-transmitting substrate.

Light-transmitting substrates provided with a light absorbing coating can be
10 used as color layers on or in front of (incandescent) lamps for general lighting purposes. The substrate may comprise, for example, a colored filter made of a piece of glass, which may or may not be flat and which is designated to be placed in a trajectory of light, said light being generated by a lamp. Such an application is often used in outdoor lighting. Another example of a light-transmitting substrate is a lamp vessel that is placed over a light source of an
15 electric lamp. Such electric lamps are predominantly used as indicator lamps in vehicles, for example as an amber-colored light source in indicators or as a red-colored light source in brake lights of automobiles. Alternative embodiments of such lamps, wherein the color temperature is increased by means of a light-absorbing coating, may also be used as headlamps of a vehicle. Said electric lamps may also be used in traffic lights.

20 An electric lamp having a lamp vessel that comprises the light-transmitting substrate according to the preamble is known from WO 01/20641 as filed by the present applicant.

The light-transmitting substrate according to WO 01/20641 is provided with an optically transparent, non-scattering, light-absorbing coating in which pigments are
25 incorporated in a sol-gel matrix and which can resist temperatures of up to 400 °C.

To manufacture light-absorbing coatings having the desired optical properties as well as having the desired thermal stability during the service life of the electric lamp, use is preferably made of inorganic pigments. In particular, the pigment is selected from the group formed by iron oxide, iron oxide doped with phosphor, zinc-iron oxide, cobalt

aluminate, neodymium oxide, bismuth vanadate, zirconium-praseodymium silicate, or mixtures thereof. Iron oxide (Fe_2O_3) is an orange pigment and P-doped Fe_2O_3 is an orange-red pigment. Zinc-iron oxide, for example ZnFe_2O_4 or $\text{ZnO} \cdot \text{ZnFe}_2\text{O}_4$ are yellow pigments. Mixing (P-doped) Fe_2O_3 with ZnFe_2O_4 yields a pigment of a deep orange color. Cobalt aluminate (CoAl_2O_4) and neodymium oxide (Nd_2O_3) are blue pigments. Bismuth vanadate (BiVO_4), also referred to as pucherite, is a yellow-green pigment. Zirconium-praseodymium silicate is a yellow pigment.

Although the above inorganic pigments do not show discoloration at high temperatures, they often incline towards having a thermochromic effect leading to a decrease in the lumen output if operated at high temperatures.

It is an object of the present invention to overcome the above drawback. Moreover, it is an object of the present invention to make transparent red, yellow, and blue coatings that are stable at high temperatures and that do not show a thermochromic effect.

To this end the present invention provides for a light-transmitting substrate according to the preamble that is characterized in that the light-absorbing particles of the light-absorbing coating comprise silver or gold or a mixture thereof, and in that the coating further comprises a dimethyl-aminosilane.

By using silver or gold in a sol-gel matrix in the presence of a dimethyl-aminosilane, transparent high temperature stable coatings can be obtained that show no thermochromic effect. The dimethyl-aminosilane acts as a stabilizer and helps controlling the particle size.

The position of the absorption maximum of the coating can be tuned by the refractive index of the matrix. In MTMS/TEOS with a refractive index of about 1.46, yellow silver-containing coatings can be made. Silver in TiO_2 or a TiO_2 /MTMS mixture can be used to make amber and, in case of an increase of the refractive index of the TiO_2 matrix, also red. Gold-containing coatings in MTMS/TEOS are red. In a TiO_2 matrix the gold containing coatings are blue.

The advantage of dimethyl-aminosilane is that dimethyl-aminosilane does not promote the sol-gel condensation reactions very strongly, whereas other aminosilanes promote the sol-gel condensation reactions too strongly. Coating liquids with an acceptable pot life can be made with the use of dimethyl-aminosilane.

Moreover, the use of dimethyl-aminosilane in the light-absorbing coating makes it possible to cure said coating at a temperature of about 350°C , which is a considerably lower temperature than the curing temperature of comparable coatings that do

not contain dimethyl-aminosilane. An advantage of such a lower curing temperature is that the substrate characteristics are not limited to a large extent. For example, the application of the coating is not restricted only to quartz glass, but ordinary soda-lime glass and simple lamp glass types can also be used as a substrate.

5 An additional benefit of the low curing temperature according to the present invention, if silver is used, is that the absorption peak for silver is very sharp. This results in a bright color of the coating layer.

 By way of comparison, reference is made to US-A-5,731,091 in which a coating is disclosed that comprises silver or gold in a sol-gel matrix in the presence of an
10 aminosilane. The specific aminosilanes that are used according to said patent are 3-aminopropyl-triethoxysilane and 3-3-(aminoethylamino)-propyl-triethoxysilane. The coatings according to US-A-5,731,091 are cured at a temperature of 500°C. Obviously, this high curing temperature does not lead to the above-mentioned advantages of the present invention. Moreover, US-A-5,731,091 discloses a yellowish-brown coating that is obtained when silver
15 is used. Such a non-bright color is due to a less sharp absorption peak.

 The dimethyl-aminosilane used advantageously comprises a dimethyl-aminopropyl-trialkoxysilane such as (N,N-dimethyl-aminopropyl)trimethoxysilane or (N,N-dimethyl-aminopropyl)triethoxysilane.

 Depending on the specific application, the substrate may comprise a specific
20 composition. In a preferred embodiment, the substrate comprises a glass substrate.

 The present invention also relates to an electric lamp comprising a light-transmitting lamp vessel which accommodates a light source, said lamp vessel comprising a light-transmitting substrate according to the above.

 As will be clear from the above, said lamp is suitable for use as an indicator
25 lamp in vehicles.

 Furthermore, the present invention provides for a method of preparing a light-absorbing coating to be applied to a light-transmitting substrate according to the above, said method at least comprising the steps of:

- 30 - preparing a hydrolysis mixture comprising a silane compound or a titanium compound that is subjected to a sol-gel process;
- dissolving a silver salt or a gold salt in an alcohol-comprising liquid and adding an dimethyl-aminosilane; and
- mixing the hydrolysis mixture and the silver or gold salt solution.

A matrix of both SiO₂ and TiO₂ may be used to incorporate the silver or gold particles.

Finally, the present invention relates to a method of applying a light-absorbing coating to a light-transmitting substrate according to the above, said method comprising the steps of:

- applying a light-absorbing coating obtained by the above method according to the invention to a light-transmitting substrate; and
- curing the light-absorbing coating at a temperature in a range of 300°C to 395°C.

The light-absorbing coating according to the present invention distinguishes itself from the prior art in that the temperature at which it can be cured can be as low as about 350°C. When the coating according to the present invention is applied to a substrate and cured in the above temperature range, a stable transparent coating is obtained which shows no thermochromic effect. Due to the fact that dimethyl-aminosilane is present, a curing temperature as low as about 350°C is sufficient.

This is also contrary to the teaching of WO 98/18736, in which curing temperatures as high as 600-900°C are mentioned. WO 98/18736 differs from the present invention in that the matrix in which the light-absorbing particles are incorporated does not comprise a sol-gel matrix. The combination of silver or gold in a sol-gel matrix in the presence of a dimethyl-aminosilane is not disclosed.

If the coating according to the present invention comprises silver, the curing is performed in a reducing atmosphere.

The present invention will be elucidated by means of the following manufacturing examples of preparing a coating and applying said coating to a substrate.

Example 1 - Gold in MTMS/TEOS

A sol-gel hydrolysis mixture is made by mixing 0.56 g ethanol, 1.63 g methyltrimethoxy silane (MTMS), 2.31 g tetraethoxy silane (TEOS), and 1.3 g 0.1 M HCl, and subjecting said mixture to hydrolysis during 4 hours. After said period 1.2 g methoxypropanol and 1.7 g water are added.

Separately, 0.3 g KAuCl_4 is dissolved in 2.2 g ethanol. After dissolution thereof, aminosilane is added in such an amount that a molar ratio gold:aminosilane of 1:2 is obtained.

5 A coating liquid is prepared by mixing the gold solution and the sol-gel hydrolysis mixture. The coating liquid is subsequently spin-coated onto the outer surface of a glass substrate. The coating is cured for 30 minutes at a temperature of 350°C , resulting in a red coating with an absorption maximum at 520 nm. The layer thickness is 1.1 μm .

Example 2 – Silver in MTMS/TEOS

10

For silver in MTMS/TEOS the method according to example 1 is followed, except that the gold salt is replaced by AgNO_3 .

AgNO_3 is dissolved in methanol in such amount that the aminosilane:Ag molar ratio is 1:1.

15

The coating liquid is prepared by mixing the silver solution and the sol-gel hydrolysis mixture. The coating liquid is subsequently spin-coated onto a glass substrate. After curing for 30 minutes at a temperature of 350°C in air not all silver had been converted. Continuation of the curing in H_2 at 350°C led to an intensively colored yellow coating with an absorption maximum at 394 nm. The layer thickness is 1.1 μm .

20

It will be clear that the present invention is not limited to the above examples. Although MTMS/TEOS is specifically mentioned as a matrix precursor for a SiO_2 matrix, it is pointed out that a TiO_2 matrix may alternatively be applied according to the present invention.

25

CLAIMS:

1. A light-transmitting substrate which is at least partly provided with a light-absorbing coating, said coating comprising light-absorbing particles that are incorporated in a sol-gel matrix, characterized in that the light-absorbing particles comprise silver or gold or a mixture thereof, and in that the coating further comprises a dimethyl-aminosilane.
- 5 2. A light-transmitting substrate as claimed in claim 1, characterized in that the dimethyl-aminosilane comprises a dimethyl-aminopropyl-trialkoxysilane.
3. A light-transmitting substrate as claimed in claim 1, characterized in that the
10 dimethyl-aminosilane comprises (N,N-dimethyl-aminopropyl)trimethoxysilane or (N,N-dimethyl-aminopropyl)triethoxysilane.
4. A light-transmitting substrate as claimed in claim 1, characterized in that the
15 substrate comprises a glass substrate.
5. An electric lamp comprising a light-transmitting lamp vessel which accommodates a light source, said lamp vessel comprising a light-transmitting substrate as claimed in any one or several of the claims 1 to 4.
- 20 6. A method of preparing a light-absorbing coating to be applied to a light-transmitting substrate as claimed in any one of the claims 1 to 4, at least comprising the steps of:
 - preparing a hydrolysis mixture comprising a silane compound or a titanium compound that is subjected to a sol-gel process;
 - 25 - dissolving a silver salt or a gold salt in an alcohol-comprising liquid and adding an dimethyl-aminosilane; and
 - mixing the hydrolysis mixture and the silver or gold salt solution.

7. A method of applying a light-absorbing coating to a light-transmitting substrate as claimed in any one of the claims 1 to 4, said method comprising the steps of:

- applying a light-absorbing coating obtained in accordance with to claim 6 to a light-transmitting substrate; and

5 - curing the light-absorbing coating at a temperature in a range of 300°C to 395°C.

8. A method as claimed in claim 7, characterized in that the curing step is performed at a temperature of 350°C.

10 9. A method as claimed in claim 7, characterized in that the coating comprises silver, and in that the curing is performed in a reducing atmosphere.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/IB 03/00084

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 H01J61/40 H01K1/32 G02B5/20 G02B5/22 C09D183/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 H01J H01K G02B C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5 731 091 A (JONSCHKER GERHARD ET AL) 24 March 1998 (1998-03-24) claims 1-8,10,12,13 column 1, line 4 - line 6 column 1, line 45 - line 53 column 1, line 61 - line 62 column 3, line 9 - line 15 column 6; example 8 ---	1-8
X	US 6 156 388 A (LINDENSTRUTH MARION ET AL) 5 December 2000 (2000-12-05) claims 1-14 column 2, line 34 - line 38 column 2, line 60 - column 3, line 12 column 3, line 32 - line 35 column 3, line 42 - line 65 --- -/--	1,4-6

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents:

A document defining the general state of the art which is not considered to be of particular relevance

E earlier document but published on or after the international filing date

L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

O document referring to an oral disclosure, use, exhibition or other means

P document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

& document member of the same patent family

Date of the actual completion of the international search

24 April 2003

Date of mailing of the international search report

06/05/2003

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Depijper, R

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 01 20641 A (KONINKL PHILIPS ELECTRONICS NV) 22 March 2001 (2001-03-22) cited in the application claims 1-9 ---	1
P,A	PATENT ABSTRACTS OF JAPAN vol. 2002, no. 10, 10 October 2002 (2002-10-10) & JP 2002 161237 A (FUJI SHIKISO KK), 4 June 2002 (2002-06-04) abstract ---	1
A	PATENT ABSTRACTS OF JAPAN vol. 2000, no. 09, 13 October 2000 (2000-10-13) & JP 2000 160104 A (MITSUBOSHI BELTING LTD), 13 June 2000 (2000-06-13) abstract ---	1
A	EP 0 452 922 B (MATSUSHITA ELECTRIC IND CO LTD) 23 October 1991 (1991-10-23) claims 1-15 -----	1

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/IB 03/00084

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 5731091	A	24-03-1998	DE 4338360 A1	11-05-1995
			DE 59405731 D1	20-05-1998
			WO 9513249 A1	18-05-1995
			EP 0729442 A1	04-09-1996
			JP 9504768 T	13-05-1997
US 6156388	A	05-12-2000	DE 19645043 A1	07-05-1998
			AT 205816 T	15-10-2001
			DE 59704690 D1	25-10-2001
			DK 938457 T3	19-11-2001
			WO 9818736 A1	07-05-1998
			EP 0938457 A1	01-09-1999
			ES 2161477 T3	01-12-2001
			JP 2001511193 T	07-08-2001
WO 0120641	A	22-03-2001	CN 1321329 T	07-11-2001
			WO 0120641 A1	22-03-2001
			EP 1129470 A1	05-09-2001
			JP 2003509825 T	11-03-2003
JP 2002161237	A	04-06-2002	NONE	
JP 2000160104	A	13-06-2000	NONE	
EP 0452922	B	23-10-1991	JP 4001602 A	07-01-1992
			JP 2825926 B2	18-11-1998
			JP 4002635 A	07-01-1992
			DE 69124410 D1	13-03-1997
			DE 69124410 T2	10-07-1997
			EP 0452922 A1	23-10-1991
			US 5667888 A	16-09-1997